



TABLE FOR CALCULATING CORRECT DOSAGE OF CITROSTAB rH

Guidance table to calculate the amount of Citrostab rH needed to consume the oxygen dissolved in wine and the resulting increase in sulphur dioxide and total acidity.

Dissolved O ₂ (mg/L)	CITROSTAB minimum dosage (g/hL)	SO ₂ released (mg/L)	SO ₂ oxidised into SO ₄ (mg/L)	Approx. increase of total SO ₂ (mg/L)	Approx. increase of free SO ₂ (mg/L)	Approx. increase of total acidity (g/L)
1	6	3.4	2	1.4	0.9	0.07
2	12	6.8	4	2.8	1.9	0.13
3	18	10	6	4	2.7	0.2
4	24	13.4	8	5.4	3.6	0.26
5	30	17.3	10	7.3	4.9	0.33
6	36	20.8	12	8.8	5.9	0.4
7	42	23.5	14	9.5	6.3	0.46
8	48	26.9	16	10.9	7.3	0.53
Pinking	50	28				0.55

10 g/hL "provide" about 0.11 g/L total acidity

PRACTICAL EXAMPLE: production of sulphur freee wines (SO₂ < 10ppm)

Before body feed filtration: 15 g/hL Citrostab rH

Estimated disolved $O_2 = 2.5 \text{ ppm} \Rightarrow$ estimated consumed $SO_2 = 5 \text{ ppm} \Rightarrow$ only 3 ppm of SO_2 measurable in wine as total SO_2 and 1 ppm as free SO_2

During rack-off: 8 g/hL Citrostab rH

Estimated disolved $O_2 = 1.5$ ppm \rightarrow estimated consumed $SO_2 = 3$ ppm \rightarrow only 1.5 ppm of SO_2 measurable in wine as total SO_2 and 0.5 mg/L as free SO_2

At bottling: 10 - 25 g/hL Citrostab rH

Dosage depends on the estimated/measured quantity of O_2 dissolved during bottling. Knowing that 6 g/hL of CitrostabrH blocks about 1 ppm O_2

Example of 4 ppm O₂ intake and 25 mg/L CitrostabrH addition SO₂ addition = 12.5 mg/L \rightarrow estimated consumed SO₂ = 8 mg/L \rightarrow only 4 ppm of SO₂ measurable in wine